

## 4,4',5,5'-Tetrakis(methylsulfonyl)tetrathiafulvalene

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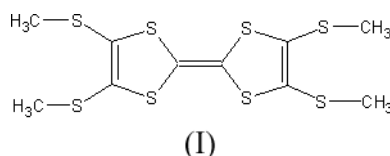
## Key indicators

Single-crystal X-ray study  
*T* = 296 K  
Mean  $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$   
*R* factor = 0.031  
*wR* factor = 0.079  
Data-to-parameter ratio = 21.6For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound,  $\text{C}_{10}\text{H}_{12}\text{S}_8$ , crystallizes in a monoclinic space group with one molecule in the asymmetric unit. All bond lengths and angles in the molecule are normal. The crystal packing is stabilized by van der Waals forces.

## Comment

Tetrathiafulvalene (TTF) and its derivatives have been studied intensively since the 1970s (Narita *et al.*, 1976; Andreu *et al.*, 2000). In the past, they were widely used in the preparation of charge-transfer salts, possessing transport properties (Williams *et al.*, 1992). Nowadays, they are also much utilized as versatile functionalized building blocks. TTF has stimulated much effort in the preparation of its derivatives, possessing various electron-donating properties.



We report here the crystal structure of the title compound, (I) (Fig. 1). All bond lengths and angles in (I) are normal (Table 1) and comparable with those in previously reported analogues (Katayama *et al.*, 1985). The crystal packing (Fig. 2) is mainly stabilized by van der Waals forces, though relatively short intermolecular  $\text{S} \cdots \text{S}$  [3.512 (2)–3.528 (2)  $\text{\AA}$ ] contacts are also observed.

## Experimental

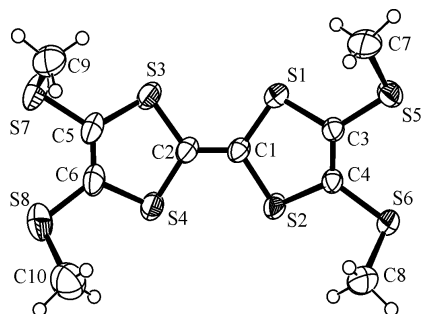
The title compound was synthesized according to the method of Tsujimoto *et al.* (1985). Crystals of (I) suitable for single-crystal X-ray diffraction were obtained by recrystallization from  $\text{CH}_2\text{Cl}_2$ .

## Crystal data

$\text{C}_{10}\text{H}_{12}\text{S}_8$	$D_x = 1.732 \text{ Mg m}^{-3}$
$M_r = 388.68$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/n$	Cell parameters from 19 853 reflections
$a = 14.0010$ (2) $\text{\AA}$	$\theta = 3.4\text{--}27.5^\circ$
$b = 7.0811$ (1) $\text{\AA}$	$\mu = 1.18 \text{ mm}^{-1}$
$c = 15.6584$ (3) $\text{\AA}$	$T = 296$ (2) K
$\beta = 106.213$ (1) $^\circ$	Block, orange
$V = 1490.67$ (4) $\text{\AA}^3$	$0.45 \times 0.35 \times 0.27 \text{ mm}$
$Z = 4$	

## Data collection

Nonius KappaCCD diffractometer	2515 reflections with $I > 2\sigma(I)$
$\varphi$ and $\omega$ scans	$R_{\text{int}} = 0.058$
Absorption correction: multi-scan (Blessing, 1995)	$\theta_{\text{max}} = 27.9^\circ$
$T_{\text{min}} = 0.620$ , $T_{\text{max}} = 0.742$	$h = -18 \rightarrow 18$
33 274 measured reflections	$k = -9 \rightarrow 9$
3521 independent reflections	$l = -20 \rightarrow 20$



**Figure 1**  
A view of (I), showing displacement ellipsoids at the 50% probability level.

#### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.031$   
 $wR(F^2) = 0.079$   
 $S = 1.05$   
 3521 reflections  
 163 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0383P)^2 + 0.2485P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 0.38 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.38 \text{ e } \text{\AA}^{-3}$   
 Extinction correction: *SHELXL97*  
 Extinction coefficient: 0.0134 (9)

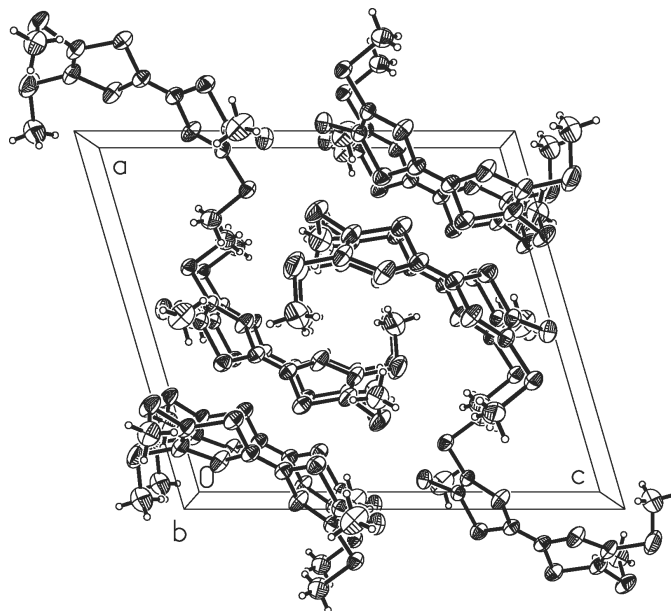
**Table 1**

Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

S1–C1	1.669 (2)	S6–C4	1.7305 (19)
S1–C3	1.763 (2)	S6–C8	1.777 (3)
S2–C1	1.678 (2)	S7–C9	1.665 (3)
S2–C4	1.703 (2)	S7–C5	1.697 (2)
S3–C2	1.687 (2)	S8–C6	1.727 (2)
S3–C5	1.760 (2)	S8–C10	1.743 (3)
S4–C2	1.672 (2)	C1–C2	1.348 (3)
S4–C6	1.720 (2)	C3–C4	1.255 (3)
S5–C7	1.671 (2)	C5–C6	1.261 (3)
S5–C3	1.6849 (19)		
C1–S1–C3	95.30 (9)	S4–C2–S3	110.02 (12)
C1–S2–C4	99.15 (10)	C4–C3–S5	121.42 (16)
C2–S3–C5	95.23 (11)	C4–C3–S1	117.39 (16)
C2–S4–C6	98.23 (11)	S5–C3–S1	120.75 (11)
C7–S5–C3	96.58 (12)	C3–C4–S2	113.43 (16)
C4–S6–C8	100.45 (11)	C3–C4–S6	123.64 (16)
C9–S7–C5	97.78 (12)	S2–C4–S6	122.38 (12)
C6–S8–C10	102.90 (13)	C6–C5–S7	122.4 (2)
C2–C1–S1	125.86 (16)	C6–C5–S3	116.81 (18)
C2–C1–S2	124.06 (17)	S7–C5–S3	120.52 (15)
S1–C1–S2	110.09 (12)	C5–C6–S4	113.89 (19)
C1–C2–S4	123.30 (16)	C5–C6–S8	124.03 (19)
C1–C2–S3	126.68 (17)	S4–C6–S8	121.91 (15)

All H atoms were positioned geometrically and refined as riding atoms, with C–H distances of 0.96  $\text{\AA}$  and  $U_{\text{iso}}(\text{H})$  values of  $1.5U_{\text{eq}}(\text{C})$ .

Data collection: *COLLECT* (Nonius, 1997–2000); cell refinement: *HKL SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *HKL DENZO* (Otwinowski & Minor, 1997) and *SCALEPACK*;



**Figure 2**  
The crystal packing of (I), viewed along the  $b$  axis.

program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *XP* (Bruker, 1998); software used to prepare material for publication: *SHELXL97*.

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